SYNTHESIS OF NEW HETEROCYCLIC PHENOLS : 9-HYDROXY-PYRIDO [1,2-a] PYRIMIDINE-4- ONE AND 9-HYDROXY-PYRIMIDO [1,6-a] PYRIMIDINE-4-ONE

F. DENNIN, D. BLONDEAU and H. SLIWA^{*}

Laboratoire de Chimie Organique, Université des Sciences et Techniques de Lille Flandres Artois, 59655 VILLENEUVE D'ASCQ, France.

Summary : The novel title phenols have been prepared by condensation of a derivative of Meldrum acid with 3-benzyloxy-2-amino-pyridine or 5-benzyloxy-4-amino-pyrimidine, and subsequent hydrogenolysis of the protecting group.

We have previously outlined the interest deserved to heterocyclic "phenols" both for their physicochemical properties (zwitterionic tautomers¹, complexing properties²) and for their enhanced reactivity³. Pursuing our work on fused heterocycles derived from 3-hydroxy-pyridine⁴ and 5-hydroxy-pyrimidine⁵ we have now extended the field of our investigation to the synthesis of 9-hydroxy-pyrido [1,2-a] pyrimidine-4-one <u>1</u> and 9-hydroxy-pyrimido [1,6-a] pyrimidine-4-one <u>2</u> which is reported here.

Condensation of 3-benzyloxy-2-amino-pyridine 3 prepared according to Bristol <u>et al</u>⁶, with diethyl ethoxymethylenemalonate yielded first the condensation product 4 (97 %) which readily underwent cyclisation to the pyrido [1,2-a] pyrimidone 5 (85 %). Obviously the presence



of the benzyloxy group prevents the rearrangement to the pyrido [2,3-b] pyridine 6 to take place as it is usually observed upon heating in the case where R#H and $X=H^7$.

Subsequent hydrogenolysis of the benzvl protecting group provided 3-ethoxycarbonyl-9-hydroxy-pyrido [1,2-a] pyrimidine-4-one 7 $(25 \%)^8$. Saponification of the compound 5 or 7 and subsequent decarboxylation were unsuccessfull to obtain the expected product 1. In all cases, we observe a ring clivage in basic media, leading to an N-aryl amino acrylic acid from which the starting product 3-benzyloxy-2-amino-pyridine was recovered.

Substituting diethyl ethoxymethylenemalonate for isopropylidene methoxymethylenemalonate 8 (prepared in situ by condensation of methyl formate with Meldrum acid in presence of zinc chloride)⁹, we could obtain by condensation with 2-amino-3 -benzyloxy-pyridine the isopropylidene N-(3-benzyloxy-2-pyridyl) aminomethylenemalonate 9 (68 %). Further heating at 250° (47 %) and subsequent hydrogenolysis of the benzyl group provided 9-hydroxy-pyride [1, 2-a] $(30\%)^8$. The above pyrimidine-4-one 1 synthesis could be extended to 5-benzyloxy-4-amino-pyrimidine 10^{10} which reacted with identical reagents under similar conditions, afforded the final 9-hydroxy-pyrimido [1,6-a] pyrimidine-4-one 2.



All the new compounds described in this report show the expected spectral properties in UV, IR and 1 H NMR and gave satisfactory mass spectrum and elemental analysis.

Further syntheses involving replacement of isopropylidene methoxy methylene malonate by 1,1,3,3-tetramethoxypropan and ring closure in perchloric acid¹¹ as well as extension of these condensations to pyrazine are under investigation.

References and Notes

- 1. R. RYDZKOWSKI, D. BLONDEAU, H. SLIWA and C. CAZE, J. Chem. Research, 1986, S, 50, M, 670. P. DECOCK, B. DUBOIS, D. BLONDEAU, H. SLIWA and H. KOZLOWSKI, J. Coord. Chem., 1988, in 2.
- press.
- 3. R. RYDZKOWSKI, D. BLONDEAU, H. SLIWA, C. CAZE, C. CERF, J. Chem. Research, 1988, S, 324, м, 2474.
- a. R. RYDZKOWSKI, D. BLONDEAU and H. SLIWA, Tetrahedron Letters, 1985, 2571. 4.
- b. R. RYDZKOWSKI, D. BLONDEAU and H. SLIWA, J. Chem. Research, 1986, S, 404, M, 3368.
- a. O. ROUSSEAUX, D. BLONDEAU and H. SLIWA, <u>Tetrahedron Letters</u>, 1986, <u>27</u>, 3127. b. O. ROUSSEAUX, D. BLONDEAU and H. SLIWA, J. Heterocyclic Chem., 1988, in press. 5.
- J. A. BRISTOL, I. GROSS and R.C. LOVEY, Synthesis, 1981, 971.
 G. R. LAPPIN, J. Am. Chem. Soc., 1948, 70, 3348.
 Yields of the debenzylation step have not been optimized.

- O.H. ABDIRIZAK, I. HERMECZ and Z. MESZAROS, Magyar Kemiai Folyrirat, 1978, 84, 494.
 a. J.H. CHESTERFIELD, J.F.W. MAC OMIE and M.S. Tute, J. Chem. Soc., 1960, 4590.
 b. J.F.W. MAC OMIE and A.B. TURNER, J. Chem. Soc., 1963, 5590.
- 11. R.H. SAWYER and D.G. WIBBERLEY, J. Chem. Soc. Perkin I, 1973, 1138.

(Received in France 28 December 1988)